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A Divergent Synthesis of (+)-Muscarine and (+)-*epi*-Muscarine from D-Glucose

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Abstract—A novel stereospecific synthesis of (+)-muscarine and (+)-*epi*-muscarine has been achieved by utilizing D-glucose as a chiral precursor. The key steps of the synthesis involved stereospecific cyclization of 3,5-di-*O*-sulfonyl-D-glucofuranose derivatives into the corresponding 2,5-anhydrides, and stereospecific hydrogenation of 2,5-anhydro-L-*threo*-hex-2-enose ethylene acetal derivatives, thus providing an access to divergent intermediates for the preparation of both target molecules in a fully stereospecific manner. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

(+)-Muscarine (1; Fig. 1) is a principal alkaloid of the poisonous mushroom *Amanita muscaria*, which shows a strong and specific cholinomimetic activity.¹ Consequently, its structure, chemistry and biological activity have been extensively studied.² There is a renewed interest in the muscarinic field due to the discovery of a relationship between cholinergic deficits and the pathology of Alzheimer's disease.³ Hence, synthetic activity in this area has been considerable, and numerous syntheses of muscarine⁴⁻⁹ and of many of its analogs^{10,11} have been accomplished from different precursors. Major drawbacks of most of these approaches are either lack of selectivity or the usage of relatively expensive reagents and/or starting compounds. Apart from a recent synthesis of (–)-muscarine from *S*-malic acid,⁹ none of the reported routes are suitable for the preparation of 5-substituted muscarine analogs.

In the course of our recent studies related to the preparation of enantiomerically pure muscarine stereoisomers by chirality transfer from D-glucose, the syntheses of (+)*epiallo*-muscarine¹² (**4**) and (-)-*allo*-muscarine¹³ (**3**) were already completed. Herein we report a divergent synthesis of (+)-muscarine (**1**) and (+)-*epi*-muscarine (**2**) based on D-glucose as a chiral precursor.¹⁴

Results and Discussion

The key steps in the synthesis of both targets **1** and **2** are: (*i*) the formation of the 2,5-anhydro-L-idose ethylene acetal derivatives **8** and **9** (Scheme 1) by an intramolecular $S_N 2$ process which is expected to occur during an acid catalyzed alcoholysis of the protected furanoses^{12,15} **6** and **7**; and (*ii*) a stereoselective catalytic reduction of the conformationally constrained dihydrofurans **12** and **15** which should be



Figure 1. (+)-Muscarine and its biologically active stereoisomers.

Keywords: 2,5-anhydro sugars; D-glucose; (+)-muscarine; (+)-epi-muscarine; stereospecific synthesis.

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Scheme 1. (a) TrCl, Py, rt, 3 days, then MsCl, +4°C, 24 h, 100%; (b) TrCl, Py, rt, 3 days, then TsCl, rt 10 days, 100%; (c) ethylene glycol, TsOH, 80°C, 5 h, 53% of 8; 54% of 9; (d) Me₂C(OMe)₂, TsOH, rt, 24 h, 87% of 10, 77% of 11; (e) Bu₄NF, MeCN, N₂, reflux, 48 h, 76% of 12 from 10, 82% of 12 from 11, 67% of 15 from 13, 86% of 15 from 14; (f) PhCH(OMe)₂, TsOH, DMF, 70°C, 20 h, 60% of 13, 86% of 14; (g) H₂, PtO₂, EtOH, rt, 24 h, 94% of 16, 91% of 17.

available from the completely protected 2,5-anhydro-Lidose derivatives **10** and **11**, as well as from **13** and **14**.

For the cyclization studies both suitable protected 3,5-di-Omesyl (6) and 3,5-di-O-tosyl (7) D-glucofuranose derivatives were first prepared. Monoacetone glucose (5),¹⁶ was tritylated and subsequently mesylated in a one-pot procedure to afford the corresponding 3,5-di-O-mesyl-6-Otrityl derivative 6 in quantitative yield. Similarly, the 3,5-di-O-tosyl derivative 7 was prepared by successive treatment of 5 with trityl chloride and tosyl chloride. Both products 6 and 7 were isolated in pure form (TLC, ¹H and ¹³C NMR) after the usual workup and used in the next step without further purification. Thus, treatment of crude 6 with ethylene glycol in the presence of toluene-4-sulfonic acid as a catalyst gave the 2,5-anhydro-L-idose derivative 8 (53% from 5), while the 3,5-di-O-tosyl ester 7, under the same reaction conditions, furnished the corresponding 3-O-tosyl derivative 9 in 54% overall yield.

In the next steps studies were performed with both isopropylidene and benzylidene protection. Thus the reaction of **8** with 2,2'-dimethoxypropane under the toluene-4sulfonic acid catalysis afforded the expected 4,6-O-isopropylidene derivative **10** in 87% yield. The 3-O-tosyl ester **9** under the same reaction conditions gave 77% yield of **11**. Both 3-sulfonates **10** and **11** readily reacted with tetrabutylammonium fluoride, in boiling acetonitrile, to afford the corresponding 2,3-unsaturated derivative **12** in

76 and 82% yield, respectively. Condensation of 8 with α, α' -dimethoxytoluene in DMF, in the presence of catalytic amounts of toluene-4-sulfonic acid, gave the corresponding 4,6-O-benzylidene derivative **13** (60%), which was subsequently treated with tetrabutylammonium fluoride to yield the olefin 15 (40% from 8). However, successive treatment of 9 with α, α' -dimethoxytoluene and tetrabutylammonium fluoride led to the formation of 15 with considerably better overall yield (74% from 9). Catalytic hydrogenation of both 12 and 15 (PtO₂, EtOH) took place stereospecifically, from the less hindered β -face, allowing the isolation of the corresponding 3-deoxy derivatives 16 (94%) and 17 (91%) as the only stereoisomers. The stereochemistry of 17 was unambiguously confirmed by NOE differential ¹H NMR spectroscopy, and the characteristic NOE relations are shown in Scheme 1. Upon irradiation of the multiplet at 4.48 ppm (2H, H-4 and H-6b), a significantly stronger NOE was observed with H-3b (2.35 ppm) than with H-3a (2.27 ppm). This result is consistent with a *cis* arrangement of H-4 and H-3b as well as with a *trans* relationship of H-4 and H-3a. However, an irradiation of H-1 (5.12 ppm) gave a strong NOE with H-3a thus proving a spatial vicinity of these protons, and consequently an α -orientation of the dioxolane acetal ring. Finally, the large vicinal coupling between H-2 and H-3b ($J_{2,3b}=9.2$ Hz) that is compatible with *cis* relationship of these protons additionally confirmed the L-xylo configuration of the molecule 17. Compared to the target 2, both intermediates 16 and 17 have the correct stereochemistry at all chiral centers. For the sake of better



Scheme 2. (a) NBS, BaCO₃, CCl₄, N₂, reflux, 1.5 h, 63% of 18, 66% of 19.

functional resemblance to the target, they must be further subjected to a C-6 deoxygenation process. This would lead to the 6-deoxy derivative 22 (Scheme 2), a potential divergent intermediate for the synthesis of both targets 1 and 2. At first it has been planned to deoxygenate the C-6 via the 6-bromodeoxy derivative 17a. It was further assumed that the compound 17 could be converted to 17a by using the well known Hanessian–Hullar reaction, which was successfully used for the conversion of numbered 4,6-*O*-benzylidene sugar acetals into the corresponding 4-*O*-benzoyl-6-bromo-6-deoxy derivatives.¹⁷ However, when the recommended reaction conditions were applied to the 3-deoxy-L-xylo-hexose 17 (NBS, BaCO₃, CCl₄ \downarrow),¹⁷ the 4-bromodeoxy derivative 18 was unexpectedly formed in 63% yield; no traces of the expected 6-bromodeoxy derivative 17a was observed. Conversely, L-ido derivative 13a,¹⁸ under the same reaction conditions, gave the expected 6-bromodeoxy derivative 19 as the only reaction product in 66% yield. The same reaction course was

observed upon treatment of 3-*O*-mesyl derivative **13** with NBS in boiling tetrachloromethane.¹⁸

The difference in product distribution between the L-xylo and L-ido series may be due to different steric and presumably electronic effects. According to a proposed mechanism¹⁹ of the process, the initial attack of a free radical at the benzylic hydrogen atom in both molecules 17 and 13a would occur first. The resulting bromoacetals (17b and 13b) could further collapse to the cyclic benzoxonium ions (18a and 19a) and bromide anion. The reaction would then assume ionic character and the more-susceptible carbon atom would be attacked preferentially by bromide ion to give the corresponding O-benzoylated bromohydrin. It seems that the preferential nucleophilic attack at C-4 in the intermediate 18a (leading to the formation of 18) is due to the overcrowding of the primary center by the synoriented dioxolane acetal function. On the contrary, the C-4 in **19a** is presumably more crowded by the β -oriented



Figure 2. Stereochemical relationships in the optimized structures 18a and 19a. The numbers in parentheses denote values of formal charge at C-4 and C-6, respectively.



Scheme 3. (a) TFA, MeOH, rt, 0.5 h, 79%; (b) AcOH, H₂O, reflux, 7 h, 81%; (c) H₂, 10% Pd/C, EtOH, AcOH, rt, 16 h, 83%; (d) TsCl, Py, -28° C, 6 days, 80%; (e) LiAlH₄ THF, N₂, reflux, 4 h 90%; (f) TsCl, Py, rt, 48 h, 80%; (g) KOBz, DMF, 100°C, 24 h, 66%; (h) DEAD, PhCO₂H, Ph₃P, THF, 0°-rt, 20 h; (i) TFA, 6 M HCl, +4°C, 24 h; (j) NaBH₄, MeOH, rt, 2 h, 60% from **24**, 27% from **22**; (k) imidazole, Ph₃P, I₂, toluene, N₂, reflux, 3 h, 84%; (l) K₂CO₃, MeOH, THF, rt, 1.5 h, 83%; (m) Me₃N, EtOH, 80°C, 3 h, 93%.

dioxolane functionality, thus directing the nucleophilic attack towards the less-hindered C-6, whereupon the observed product **19** was formed. These assumptions were verified by molecular modeling studies.²⁰

Preliminary molecular mechanics calculations (MM+) gave the low energy conformations of 18a and 19a with the tetrahydrofuran rings having the ${}^{4}T_{3}$ and the ${}_{4}T^{3}$ geometry, respectively (Fig. 2). These findings certainly do not exclude the existence of the other conformations, but apparently suggest that both intermediates 18a and 19a may occupy conformations suitable to explain the experimental results. Indeed, a careful examination of the optimized structure 18a clearly indicated that the dioxolane acetal function does prevent approach of the nucleophile to C-6. On the contrary, the spatial arrangement of both C-2 and C-3 substituents in 19a obviously causes serious overcrowding of the C-4. Moreover, semiempirical PM3 calculations performed on 18a gave significantly different values for the formal charge at C-4 (0.050) and C-6 (-0.039), thus indicating a higher electrophilicity of C-4 in the intermediate 18a. Conversely, concerning the calculated formal charges in 19a, the C-6 (0.052) was shown to be somewhat more electrophilic than the C-4 (0.020). These findings appear to convincingly explain the experimental results.

Due to the undesirable outcome of the last reaction, an alternative procedure for introduction of 6-deoxy functions

into the molecules **16** and **17** has been developed. Selective removal of the 4,6-*O*-isopropylidene protective group in **16**, achieved with 10% trifluoroacetic acid in methanol, gave the corresponding diol **20** in 79% yield, while the action of diluted acetic acid onto **17** afforded 81% yield of **20** (Scheme 3). Finally, the intermediate **20** was more conveniently prepared directly from **15** in 83% yield, by a one-pot procedure which included a catalytic hydrogenation of the double bond, and a hydrogenolytic removal of the benzylidene protection over 10% Pd/C. Monotosylation of the diol **20** at -28° C produced 6-*O*-tosyl derivative **21** (80%) which was subsequently treated with lithium aluminum hydride in boiling tetrahydrofuran, to give the key chiral intermediate **22** in 90% yield.

The seven-step synthetic sequence, which uses the tosyloxy leaving group as well as the benzylidene protection, obviously represents the most convenient route towards the divergent intermediate **22**, since it provided the highest overall yield of the desired product (24% from **5**). Reaction of **22** with tosyl chloride in pyridine gave the corresponding 4-*O*-tosyl derivative **23** in 80% yield. Compound **23** readily reacted with potassium benzoate, to give the chiral synthon **24** (66%) with an absolute configuration of all stereocenters corresponding to (+)-muscarine (1). Compound **24** was alternatively prepared directly from **22** by using the standard Mitsunobu conditions.²¹ However, thus obtained sample **24** was slightly contaminated with unidentified aromatic impurities that remained in the sample even after repeated



Scheme 4. (a) BzCl, Py, rt, 24 h, 86%; (b) TFA, 6 M HCl, $+4^{\circ}$ C, 24 h; (c) NaBH₄, MeOH, rt, 2 h, 59% from **29**; (d) imidazole, Ph₃P, I₂, toluene, N₂, reflux, 3 h, 90%; (e) K₂CO₃, MeOH, THF, rt, 1.5 h, 65%; (f) Me₃N, EtOH, 80°C, 3 h, 95%.

chromatographic purification. Fortunately, these impurities did not affect the course of the following reaction directed to the hydrolytic removal of the dioxolane protective group. Thus, treatment of 24 with a 4:1 mixture of trifluoroacetic and 6 M hydrochloric acid gave the unstable aldehyde 25, which was immediately reduced with sodium borohydride to afford the primary alcohol 26. It appeared that the fourstep sequence realized via the 4-O-tosyl derivative 23 represents a slightly more convenient procedure for the preparation of 26, since it provided a somewhat higher overall yield (32% from 22) compared to the three-step sequence based on Mitsunobu reaction (27% from 22). Reaction of 26 with iodine, imidazole and triphenylphosphine, according to the methodology developed by Garegg and Samuelsson,²² gave the known⁷ iodo derivative **27** in 84% yield. *O*-Debenzoylation of 27 with potassium carbonate in methanol afforded 83% yield of the corresponding iodo alcohol 28. Finally, compound 28 was converted to (+)-muscarine iodide (1) by treatment with trimethylamine in ethanol. Conversion of the intermediate 22 into the (+)-epi-muscarine iodide (2) is outlined in Scheme 4.

Table 1. ¹H and ¹³C NMR spectral data for 1 and 2 (in D₂O)

The sequence started with O-benzoylation of 22 whereupon the corresponding 4-O-benzoyl derivative 29 was obtained in 86% yield. Hydrolytic removal of the dioxolane protective group in 29, followed by subsequent sodium borohydride reduction of the unstable aldehyde 30 gave the corresponding primary alcohol 31 in 59% yield (51% from 22). Alternatively, when the last three-step process was carried out without chromatographic purification of the 4-O-benzoyl derivative 29, the desired product 31 was obtained in an overall yield of 53% with respect to compound 22. The intermediate 31 was finally converted to the (+)-epi-muscarine iodide (2) by using the same three-step sequence which was already applied for the conversion of 26 into the (+)-muscarine iodide (1; Scheme 3). The ¹H and ¹³C NMR spectral data (Table 1) as well as physical constants of both 1 and 2 thus obtained were in reasonable agreement with those already reported.^{7,8}

In conclusion, this paper describes a divergent 11-step synthesis of (+)-muscarine (1) and (+)-*epi*-muscarine (2) from commercially available monoacetone glucose (5) in 5

Comp.	Chemical shift (δ) and J (Hz)									Reference
	H-1a	H-1b	H-2	H-3a	H-3b	H-4	H-5	H-6		
1	3.48 3.39	3.62 3.49	4.66 4.57	2.01 1.91	2.12 2.01	4.13 4.03	4.06 3.96	1.21 1.11		This work 7
2	3.56 3.42	3.60 3.46	4.46 4.37	1.60 1.51	2.61 2.51	4.24 4.14	3.95 3.85	1.22 1.13		This work 7
	$J_{1\mathrm{a,lb}}$	$J_{1a,2}$	$J_{ m 1b,2}$	$J_{2,3a}$	$J_{2,3\mathrm{b}}$	$J_{3a,3b}$	$J_{3\mathrm{a},4}$	$J_{ m 3b,4}$	$J_{4,5}$	
1	14.0 14.0	9.1 9.2	1.8 1.8	9.5 9.6	6.4 6.3	13.7 13.7	5.9 5.7	2.4 2.3	2.5 2.5	This work 7
2	13.9 13.9	3.1 3.3	9.7 8.2	5.9 5.7	8.7 8.7	14.3 14.3	1.9 1.9	6.0 6.0	3.5 3.5	This work 7
	C-1	C-2	C-3	C-4	C-5	C-6	NMe ₃			
1	73.21 73.51	74.66 74.85	40.23 40.53	77.87 78.13	86.77 86.91	21.89 22.11	56.70 57.09			This work 7
2	72.96 73.00	73.99 73.95	41.97 41.96	74.16 74.16	83.45 83.43	16.18 16.15	56.84 56.82			This work 7

and 7% overall yield, respectively. Although this new synthesis of **1** consists of more synthetic steps and has a lower overall yield than the earlier preparation from L-rhamnose (10% from seven steps),⁵ it uses a less expensive and readily available starting material.²³ Compared with the previous carbohydrate based approaches⁴ (each containing at least one non-specific synthetic step), the new synthesis of both targets **1** and **2** from D-glucose was realized in a fully regio- and stereospecific manner. Moreover, appropriate C-6 modifications of the 2,5-anhydro derivatives of type **20** or **21**, may provide access to potential divergent intermediates for the preparation of a variety of 5-substituted (+)-muscarine analogs.

Experimental

General methods

Melting points were determined on a Büchi 510 apparatus and were not corrected. Optical rotations were measured on a Perkin-Elmer 141 MC polarimeter. IR spectra were recorded with a Specord 75IR spectrophotometer. NMR spectra were recorded on a Bruker AC 250 E instrument and chemical shifts are expressed in ppm downfield from tetramethylsilane. Mass spectra were recorded on Finnigan-MAT 8230 and VG AutoSpec mass spectrometers. TLC was performed on DC Alufolien Kieselgel 60 F₂₅₄ (E. Merck). Column chromatography was carried out using Kieselgel 60 (under 0.063 mm; E. Merck). Flash column chromatography was performed using ICN silica 32-63. All organic extracts were dried with anhydrous Na₂SO₄. Organic solutions were concentrated in a rotary evaporator under diminished pressure at a bath temperature below 35°C.

2,5-Anhydro-3-O-methanesulfonyl-L-idose ethylene acetal (8). To a solution of 5 (5.0 g, 27.22 mmol) in dry pyridine (50 mL) was added trietyl chloride (10.12 g, 36.32 mmol) and the mixture was kept at room temperature for 3 days. After cooling to 0°C, mesyl chloride (4.5 mL, 57.75 mmol) was added dropwise to the stirred and cooled solution. The mixture was left at $+4^{\circ}$ C for 24 h, then poured onto ice (80 g) and acidified with 6 M HCl (120 mL). The separated precipitate was dissolved in CH₂Cl₂ (200 mL) and the solution was washed successively with cold 10% HCl (200 mL) and water (3×100 mL). Organic layer was dried and evaporated to give chromatographically homogenous 6 $(14.04 \text{ g}, \sim 100\%)$ as a white amorphous powder: mp 170– 171°C (decomp.); R_f 0.35 (CH₂Cl₂), R_f 0.46 (9;1 toluene-EtOAc); ¹H NMR (CDCl₃): δ 1.34 and 1.51 (2×s, 3H each, CMe₂), 2.90 and 3.19 (2×s, 3H each, 2×MeSO₂), 3.47 (dd, 1H, J_{6a,6b}=11.3 Hz, J_{5,6a}=4.7 Hz, H-6a), 3.65 (dd, 1H, $J_{5,6b}=2.1$ Hz, H-6b), 3.70 (dd, 1H, $J_{3,4}=2.7$ Hz, $J_{4,5}=$ 9.4 Hz, H-4), 5.02 (d, 1H, H-3), 5.97 (d, 1H, H-1), 7.21– 7.51 (m, 15H, CPh₃). ¹³C NMR (CDCl₃): δ 26.16 and 26.43 (CMe₂), 38.62 and 39.26 (2×MeSO₂), 62.36 (C-6), 75.82 (C-4), 76.35 (C-5), 80.26 (C-3), 82.80 (C-2), 87.66 (CPh₃), 104.75 (C-1), 112.74 (CMe₂), 127.23, 127.90, 128.59 and 143.02 (CP h_3). To a suspension of crude 6 (3.79 g, 6.12 mmol) in ethylene glycol (22 mL) was added toluene-4-sulfonic acid (0.379 g, 1.99 mmol) and the mixture was stirred at 80°C for 5 h. The reaction mixture

was left at $+4^{\circ}$ overnight, then filtered through a Celite pad, and the precipitate washed with ethylene glycol. Combined filtrate and washings were poured into saturated NaCl solution (44 mL), neutralized with NaHCO₃, and extracted with EtOAc (6×20 mL). The extracts were combined, dried and evaporated to pale yellow oil. Flash column chromatography (EtOAc) of the residue afforded pure 8 (0.931 g, 53% from 5) as a colorless syrup: $[\alpha]_{D} = +29.6$ (c, 0.19 in CHCl₃); *R*_f 0.24 (19:1 CH₂Cl₂–MeOH), *R*_f 0.27 (EtOAc); $\nu_{\rm max}$ (film): 3430 (broad), 2940, 2880, 1370, 1200, 1100, 930 cm⁻¹; ¹H NMR (CDCl₃): δ 2.72 (bs, 1H, exchangeable with D₂O, OH-6), 3.11 (s, 3H, MeSO₂), 3.84-4.15 (m, 6H, 2×H-6 and CH₂-dioxolane), 4.24 (dd, 1H, $J_{1,2}$ =6.3 Hz, $J_{2,3}$ =3.9 Hz, H-2), 4.28 (m, 1H, $J_{4,5}$ =4.0 Hz, H-5), 4.38 (d, 1H, exchangeable with D_2O , $J_{4,OH}$ =4.1 Hz, OH-4), 4.64 (m, 1H, J_{3,4}=2.0 Hz, H-4), 5.01 (dd, 1H, H-3), 5.11 (d, 1H, H-1); ¹³C NMR (CDCl₃): δ 38.10 (MeSO₂), 61.53 (C-6), 65.24 and 65.46 (2×CH₂-dioxolane), 77.51 (C-4), 79.46 (C-2), 79.52 (C-5), 85.43 (C-3), 101.91 (C-1); CI-MS: m/z 569 (2M⁺+1), 285 (M⁺+1). HR-MS: Calcd for C₉H₁₆O₈S: 284.0566. Found: *m/z* 284.0571.

2,5-Anhydro-3-O-p-toluenesulfonyl-L-idose ethylene acetal (9). Compound 5 (7.5 g, 34.06 mmol) was dissolved in anhydrous pyridine (75 mL), and trityl chloride (15.0 g, 53.8 mmol) was added. After 3 days at room temperature, to solution was added tosyl chloride (20.25 g, the 106.22 mmol) and the mixture was left at room temperature for 10 days. The mixture was poured into a stirred and cooled (0°C) solution of 6 M HCl (300 mL), whereupon a white precipitate was formed. The precipitate was collected by filtration, washed with cold water and dried at 35°C for 48 h. Chromatographically homogenous 7 (26.2 g, 100%) was thus obtained as a white amorphous solid: mp 128-129°C (decomp.); $R_{\rm f}$ 0.86 (2:1 EtOAc-light petroleum), $R_{\rm f}$ 0.65 (7:1 toluene–Me₂CO): ¹H NMR (CDCl₃): δ 1.27 and 1.42 (2×s, 3H, each, CMe₂), 2.43 and 2.48 (2×s, 3H, each 2×MeC₆H₄SO₂), 3.22 (dd, 1H, J_{6a,6b}=11.0 Hz, J_{5,6a}=2.8 Hz, H-6a), 3.33 (dd, 1H, J_{5,6b}=5.5 Hz, H-6b), 4.55 (dd, 1H, J_{3,4}=2.8 Hz, J_{4,5}=5.2 Hz, H-4), 4.75 (ddd, 1H, H-5), 4.79 (d, 1H, $J_{1,2}$ =3.4 Hz, H-2), 4.93 (d, 1H, H-3), 5.71 (d, 1H, H-1), 7.41–7.86 (m, 23H, CPh₃ and 2×Me $C_6H_4SO_2$); ¹³C NMR (CDCl₃): δ 21.75 and 21.82 (2×MeC₆H₄SO₂), 26.33 and 26.55 (CMe2), 62.26 (C-6), 77.11 (C-5), 77.89 (C-4), 81.22 (C-3), 82.38 (C-2), 86.75 (CPh₃), 104.39 (C-1), 112.7 (CMe_2) , 127.06–145.5 $(CPh_3 \text{ and } 2 \times MeC_6H_4SO_2)$. A mixture of crude 7 (26.0 g) and toluene-4-sulfonic acid (1.5 g, 7.89 mmol) in ethylene glycol (150 mL) was stirred at 80°C for 2 h. After the usual workup (see preparation of 8), crude 9 remained as a crystalline solid. Recrystallization of the residue from toluene gave pure 9 (5.95 g) as white needles, mp 144-146°C. An additional amount of pure 9 (0.7 g) was obtained after purification of the mother liquor by flash column chromatography (2:1 EtOAc-light petroleum). Total yield of 9 was 6.65 g (54% from 5). Recrystallization from toluene afforded an analytical sample 9: mp 146°C; $[\alpha]_{\rm D} = +20.7$ (c, 0.68 in CHCl₃); $R_{\rm f}$ 0.17 (2:1 EtOAc-light petroleum), R_f 0.13 (3:1 toluene-Me₂CO); $\nu_{\rm max}$ (KBr): 3450–3260 (broad), 2970, 2920, 1610, 1360, 1190, 1100, 1060, 920 cm⁻¹; ¹H NMR (CDCl₃): δ 2.46 (s, 3H, $MeC_5H_4SO_2$), 2.73 (t, 1H, exchangeable with D_2O_2 , J_{6,OH}=4.0 Hz, OH-6), 3.82-3.97 (m, 5H, 2×CH₂-dioxlane and H-6a), 4.03 (dd, 1H, $J_{6a,6b}=12.7$ Hz, $J_{5,6b}=3.8$ Hz, H-6b), 4.14 (dd, 1H, $J_{1,2}$ =6.1 Hz, $J_{2,3}$ =4.2 Hz, H-2), 4.22 (m, 1H, $J_{4,5}$ =4.2 Hz, $J_{5,6a}$ =2.7 Hz, H-5), 4.30 (d, 1H, exchangeable with D₂O, $J_{4,OH}$ =4.3 Hz, OH-4), 4.55 (dd, 1H, $J_{3,4}$ =2.0 Hz, H-4), 4.91 (dd, 1H, H-3), 4.95 (d, 1H, H-1), 7.31–7.87 (m, 4H, MeC₆H₄SO₂). ¹³C NMR (CDCl₃): δ 21.76 (*Me*C₆H₄SO₂), 61.66 (C-6), 65.21 and 65.35 (2×CH₂-dioxolane), 77.09 (C-4), 79.44 (C-2), 79.52 (C-5), 84.98 (C-3), 101.62 (C-1), 128.23, 129.81, 130.01 and 145.24 (MeC₆H₄SO₂); EI-MS: *m*/*z* 359 (M⁺–1). Anal. Calcd for C₁₅H₂₀O₈S: C 49.99, H 5.59, S 8.90. Found: C 50.06, H 5.74, S 8.77.

2,5-Anhydro-4,6-O-isopropylindene-3-O-methanesulfonyl-L-idose ethylene acetal (10). A mixture of 8 (0.42 g, 1.51 mmol), toluene-4-sulfonic acid (0.0056 g, 0.03 mmol) and 2,2'-dimethoxypropane (7 mL) was stirred at room temperature for 24 h. The solution was poured into 10% NaCl solution (14 mL), neutralized with NaHCO₃ (0.01 g) and extracted with CH_2Cl_2 (3×10 mL). The extracts were combined, dried and evaporated. Flash chromatography (9:1 CH₂Cl₂-EtOAc) of the residue (0.52 g) yielded pure 10 (0.426 g, 87%) as a colorless syrup: $[\alpha]_D = +72.4$ (c, 0.64 in CHCl₃); R_f 0.28 (9:1 CH₂Cl₂-EtOAc), R_f 0.36 (1:1 EtOAc-light petroleum); ν_{max} (KBr): 3000, 2930, 1360, 1190, 1100, 930 cm⁻¹; ¹H NMR (CDCl₃): δ 1.33 and 1.41 (2×s, 3H each, CMe₂), 3.09 (s, 3H, MeSO₂), 3.80-4.11 (m, 6H, 2×CH₂-dioxolane and 2×H-6), 4.15 (m, 1H, $J_{5.6}$ = 2.3 Hz, J_{4.5}=2.5 Hz, H-5), 4.28 (dd, 1H, J_{2.3}=3.6 Hz, $J_{1,2}$ =7.2 Hz, H-2), 4.50 (dd, 1H, $J_{3,4}$ =1.1 Hz, H-4), 4.88 (dd, 1H, H-3), 5.10 (d, 1H, H-1); ¹³C NMR (CDCl₃) δ 19.20 and 28.42 (CMe₂), 37.94 (MeSO₂), 60.36 (C-6), 64.99 and 65.26 (2×CH₂-dioxolane), 72.66 (C-5), 74.08 (C-4), 80.19 (C-2), 84.55 (C-3), 97.53 (CMe₂), 101.85 (C-1). Attempted crystallization from MeOH gave small amount of the crystalline sample 10, mp 112-114°C (decomp.), along with variety decomposition products that remained in the mother liquor. Due to instability of the product 10, a correct microanalysis of HR-MS could not be obtained.

2,5-Anhydro-4,6-O-isopropylidene-3-O-p-toluenesulfonyl-L-idose ethylene acetal (11). Diol 9 (1.0 g, 2.78 mmol), toluene-4-sulfonic acid (0.01 g, 0.05 mmol) and 2,2'dimethoxypropane (10 mL) were stirred at room temperature for 24 h. The usual workup gave crude 11, which was purified by flash chromatography (2:1 light petroleum-EtOAc), to give colorless needles of pure 11 (0.86 g, 77%). Recrystallization from MeOH yielded an analytical sample **11**: mp 142–144°C; $[\alpha]_D = +79.1$ (c, 0.23 in CHCl₃); R_f 0.68 (2:1 EtOAc-light petroleum); ν_{max} (KBr): 3080, 3020, 2980, 2920, 2900, 1600, 1390, 1200, 1100, 930 cm⁻¹; ¹H NMR (CDCl₃): δ 1.29 and 1.35 (2×s, 3H each, CMe₂), 2.42 (s, 3H, MeC₆H₄SO₂), 3.53–4.92 (m, 4H, CH₂-dioxolane), 3.96 (m, 2H, $J_{5,6a}$ =2.8 Hz, $J_{5,6b}$ = 2.1 Hz, 2×H-6), 4.06 (m, 1H, J_{4,5}=2.8 Hz, H-5), 4.10 (dd, 1H, $J_{2,3}=3.7$ Hz, $J_{1,2}=7.0$ Hz, H-2), 4.44 (m, 1H, J_{3,4}=1.1 Hz, J_{2,4}=1.8 Hz, H-4), 4.81-4.91 (m, 2H, H-1 and H-3), 7.31-7.75 (m, 1H, MeC₆H₄SO₂); ¹³C NMR (CDCl₃): δ 19.45 and 28.42 (CMe₂), 21.71 (MeC₆H₄SO₂), 60.52 (C-6), 65.03 and 65.16 (2×CH₂-dioxolane), 72.88 (C-5), 73.67 (C-4), 80.52 (C-2), 84.05 (C-3), 97.68 (CMe₂), 101.51 (C-1), 128.34, 129.66, 132.90 and 145.18 $(MeC_6H_4SO_2)$; EI-MS: m/z 385 $(M^+ - Me)$. Anal. Calcd for

 $C_{18}H_{24}O_8S$: C 53.99, H 6.04, S 8.01. Found: C 54.34, H 6.30, S 8.36.

2,5-Anhydro-3-deoxy-4,6-*O***-isopropylidene-L***-threo***-hex-2-enose (12).** (*Procedure A*) To a solution of **10** (0.418 g, 1.29 mmol) in dry acetonitrile (5 mL) was added Bu_4NF (1.69 g, 6.45 mmol) and the mixture was refluxed in an atmosphere of N_2 for 48 h. The mixture was evaporated and the residue (1.626 g) purified by flash chromatography (19:1 toluene-Me₂CO), to afford pure **12** (0.224 g, 76%) as colorless syrup.

Procedure B: a mixture of 11 (0.82 g, 2.05 mmol), Bu₄NF (2.85 g, 10.9 mmol) in acetonitrile (15 mL) was refluxed in an atmosphere of N2 for 48 h. The workup followed by chromatographic purification according to procedure A, gave pure 12 (0.358 g, 82%) as a colorless syrup: $[\alpha]_{\rm D} = +86.3$ (c, 1.1 in CHCl₃); $R_{\rm f}$ 0.25 (19:1 toluene-Me₂CO), R_f 0.6 (2:1 toluene–Me₂CO); ν_{max} (film): 3100, $3010, 2950, 2900, 1690, 1380, 1210, 1120, 1040, 970 \text{ cm}^{-1};$ ¹H NMR (CDCl₃): δ 1.37 and 1.38 (2×s, 3H each, CMe₂), 3.87 (dd, 1H, $J_{6a,6b}$ =11.9 Hz, $J_{5,6a}$ =6.8 Hz, H-6a), 3.90-4.09 (m, 5H, 2×CH₂-dioxolane and H-6b), 4.50 (m, 1H, $J_{4,5}$ =6.6 Hz, H-5), 4.94 (dd, 1H, $J_{3,4}$ =2.8 Hz, H-4), 5.24 (d, 1H, H-3), 5.52 (s, 1H, H-1); ¹³C NMR (CDCl₃): δ 22.96 and 26.51 (CMe2), 58.97 (C-6), 65.29 and 65.33 (2×CH₂-dioxolane), 72.66 (C-4), 78.81 (C-5), 97.56 (C-1), 98.85 (CMe2), 99.47 (C-3), 160.87 (C-2). HR-MS: Calcd for C₁₁H₁₆O₅: 228.0998. Found: *m*/*z* 228.0992.

2,5-Anhydro-4,6-O-benzylidene-3-O-methanesulfonyl-Lidose (13). To a solution of 8 (0.55 g, 1.93 mmol) in dry DMF (5.5 mL) was added toluene-4-sulfonic acid (0.06 g, 0.32 mmol) and α, α' -dimethoxytoluene (1.2 mL, 7.99 mmol). The mixture was stirred at 70°C for 20 h, then neutralized with NaHCO₃ (0.2 g), evaporated and the residue was extracted with CH_2Cl_2 (2×10 mL). The extracts were combined, dried and evaporated to yellow oil (0.5 g). Flash column chromatography (1:1 EtOAc-light petroleum) gave pure 13 (0.43 g, 60%) as a colorless syrup: $[\alpha]_{\rm D}$ = +47.4 (c, 0.23 in CHCl₃); R_f 0.76 (EtOAc). $\nu_{\rm max}$ (film): 3040, 3000, 2960, 2910, 1620, 1400, 1370, 1200, 1160, 1100, 1020, 980, 920 cm⁻¹; ¹H NMR (CDCl₃): δ 3.13 (s, 3H, MeSO₂), 3.82–4.05 (m, 4H, 2×CH₂-dioxolane), 4.1 (dd, 1H, $J_{6a,6b}$ =13.1 Hz, $J_{5,6a}$ =1.8 Hz, H-6a), 4.24 (m, 1H, $J_{4,5}=2.4$ Hz, H-5), 4.38 (dd, 1H, $J_{1,2}=7.1$ Hz, $J_{2,3}=$ 3.7 Hz, H-2), 4.50 (d, 1H, H-6b), 4.74 (dd, 1H, $J_{3,4}$ = 1.0 Hz, H-4), 5.08 (d, 1H, H-3), 5.17 (d, 1H, H-1), 5.48 (s, 1H, PhCH), 7.31–7.52 (m, 5H, Ph); 13 C NMR (CDCl₃): δ 37.98 (MeSO₂), 65.06 and 65.34 (2×CH₂-dioxolane), 67.29 (C-6), 73.15 (C-5), 79.26 (C-4), 80.79 (C-2), 83.72 (C-3), 99.05 (PhCH), 101.88 (C-1), 126.11, 128.24, 129.2 and 137.24 (Ph). CI-MS: m/z 373 (M⁺+1). HR-MS: Calcd for C₁₁H₂₀O₈S: 372.0879. Found: *m*/*z* 372.0888.

2,5-Anhydro-4,6-*O***-benzylidene-3***-O***-p-toluenesulfonyl-L-idose ethylene acetal (14).** A mixture of **9** (5.5 g, 15.26 mmol), α, α' -dimethoxytoluene (9.7 mL, 64.62 mmol) and toluene-4-sulfonic acid (0.15 g, 0.79 mmol) in dry DMF (40 mL) was stirred at 70°C for 24 h. After neutralization with NaHCO₃ (0.8 g) the solvent was evaporated and the remaining crude residue was extracted with CH₂Cl₂ (2×20 mL). The combined extracts were dried and

evaporated to pale yellow syrup. Direct crystallization from MeOH yielded pure 14 (5.4 g) as colorless needles, mp 139–140°C. Flash chromatography (49:1, toluene– Me₂CO) of the mother liquor afforded an additional amount of pure 14 (0.50 g). Total yield 5.9 g (86%). An analytical sample 14 was obtained by recrystallization from MeOH: mp 139–140°C; $[\alpha]_{\rm D}$ =+64.5 (c, 0.34 in CHCl₃); R_f 0.73 (2:1 EtOAc-light petroleum); $R_{\rm f}$ 0.50 (4:1 toluene-Me₂CO). ν_{max} (KBr): 3080, 3020, 2900, 1600, 1370, 1190, 1100, 920 cm⁻¹; ¹H NMR (CDCl₃): δ 2.45 (s, 3H, MeC₆H₄SO₂), 3.58-3.97 (m, 4H, 2×CH₂-dioxolane), 4.07 (dd, 1H, J_{6a,6b}=12.8 Hz, J_{5,6a}=1.8 Hz, H-6a), 4.20 (m, 1H, $J_{4,5}$ =2.2 Hz, H-5), 4.27 (dd, 1H, $J_{1,2}$ =7.3 Hz, $J_{2,3}$ =3.4 Hz, H-2), 4.47 (d, 1H, H-6b), 4.74 (m, 1H, H-4), 4.95 (m, 2H, H-1 and H-3), 5.45 (s, 1H, PhCH), 7.3-7.8 (m, 9H, Ph and MeC₆H₄SO₂); ¹³C NMR (CDCl₃): δ 21.75 (MeC₆H₄SO₂), 65.05 and 65.17 (2×CH₂-dioxolane), 67.45 (C-6), 73.18 (C-5), 78.98 (C-4), 81.10 (C-2), 83.17 (C-3), 99.23 101.52 (C-1), 126.25–145.97 (Ph and (Ph*C*H), $MeC_6H_4SO_2$). EI-MS: m/z 447 (M⁺-1); CI-MS: m/z 449 (M^++1) . Anal. Calcd for $C_{22}H_{24}O_8S$: C 58.92, H 5.39, S 7.15. Found: C 59.13, H 5.17, S 7.36.

2,5-Anhydro-4,6-*O***-benzylidene-3-deoxy-L***-threo***-hex-2-enose ethylene acetal (15).** (*Procedure A*) A mixture of **13** (0.104 g, 0.28 mmol) and Bu₄NF (0.235 g, 0.9 mmol) in dry MeCN (2 mL) was refluxed in an atmosphere of N₂ for 48 h. The reaction mixture was diluted with CH_2Cl_2 (15 mL) and washed with water (3×10 mL). Organic solution was dried and evaporated to brown syrup (0.08 g). Chromatographic purification on a column of flash silica (7:3 light petroleum–Me₂CO) afforded pure **15** (0.052 g, 67%) as a white crystal-line solid.

Procedure B: to a solution of 3-O-tosyl derivative 14 (1.67 g, 3.72 mmol) in dry acetonitrile (20 mL) was added Bu₄NF (4.0 g, 15.29 mmol) and the mixture was refluxed in an atmosphere of N_2 for 24 h. The solvent was evaporated off; the residue was purified by flash chromatography (49:1 toluene-Me₂CO) to yield pure 15 (0.888 g, 86%) as colorless needless. Recrystallization from MeOH gave an analytical sample 15: mp 91°C; $[\alpha]_D = +130.2$ (c, 0.42 in CHCl₃); R_f 0.68 (2:1 toluene–Me₂CO). ν_{max} (KBr): 3080, 3010, 2920, 1670, 1600, 1470, 1400, 1330, 1250, 1210, 1130, 1040, 950 cm⁻¹; ¹H NMR (CDCl₃); δ 3.96–4.10 (m, 4H, 2×CH₂-dioxolane), 4.13 (m, 1H, $J_{4,5}$ =4.6 Hz, $J_{5,6a}$ =3.0 Hz, $J_{5,6b}$ =1.0 Hz, H-5), 4.29 (dd, 1H, $J_{6a,6b}$ = 13.7 Hz, H-6a), 4.65 (d, 1H, H-6b), 4.94 (dd, 1H, J_{3,4}=2.9 Hz, H-4), 5.47 (m, 2H, H-1 and H-3), 5.61 (s, 1H, PhCH), 7.25–7.45 (m, 5H, Ph); 13 C NMR (CDCl₃): δ 65.12 and 65.23 (2×CH₂-dioxolane), 66.27 (C-6), 76.32 (C-4), 77.2 (C-5), 97.58 (PhCH), 98.08 (C-3), 101.14 (C-1) 126.17, 128. 21, 128.9 and 137.94 (Ph), 162.08 (C-2). EI-MS: m/z 276 (M⁺). Anal. Calcd for C₁₅H₁₆O₅: C 65.21, H 5.84; Found: C 65.40, H 5.92.

2,5-Anhydro-3-deoxy-4,6-*O***-isopropylidene-L***-xylo***-hexose ethylene acetal (16).** A solution of **12** (0.87 g, 3.81 mmol) in EtOH (20 mL) was hydrogenated over PtO_2 (0.08 g, 0.35 mmol) for 24 h at room temperature. The mixture was filtered and the catalyst washed with EtOAc. The filtrate and washings were combined and evaporated to give crude **16** that was purified by flash column chromatography (9:1 toluene–Me₂CO). Pure **16** (0.827 g, 94%) was obtained as a colorless syrup: $[\alpha]_D = +19.7$ (*c*, 1.16 in CHCl₃); R_f 0.4 (2:1 toluene–Me₂CO). ν_{max} (film): 3000, 2970, 2920, 1390, 1290, 1120, 970 cm⁻¹; ¹H NMR (CDCl₃): δ 1.36 and 1.40 (2×s, 3H, each, CMe₂), 2.07 (dd, 1H, $J_{2,3a}=3.6$ Hz, $J_{3a,3b}=$ 14.3 Hz, H-3a), 2.28 (ddd, 1H, $J_{2,3b}=9.2$ Hz, $J_{3b,4}=5.3$ Hz, H-3b), 3.72 (m, 1H, $J_{5,6}=3.0$ Hz, $J_{4,5}=2.8$ Hz, H-5), 3.81–4.08 (m, 6H, 2×CH₂-dioxolane and 2×H-6), 4.35 (m, 1H, H-4), 5.03 (d, 1H, $J_{1,2}=7.1$ Hz, H-1); ¹³C NMR (CDCl₃): δ 19.56 and 28.38 (CMe₂), 35.38 (C-3), 60.65 (C-6), 65.09 and 65.14 (2×CH₂-dioxolane), 70.01 (C-4), 75.22 (C-5), 79.16 (C-2), 97.61 (CMe₂), 104.86 (C-1). CI-MS: *m/z* 231 (M⁺+1). HR-MS: Calcd for C₁₁H₁₈O₅: 230.1154. Found: *m/z* 230.1161.

2,5-Anhydro-4,6-O-benzylidene-3-deoxy-L-xylo-hexose ethylene acetal (17). Compound 15 (1.3 g, 4.7 mmol) in EtOH (15 mL) was hydrogenated over PtO_2 (0.13 g, 0.57 mmol) for 24 h. After workup as described above (preparation of 16), crude 17 was obtained which was purified on a column of silica gel (100 g, 5:1 toluene- Me_2CO). Pure 17 (1.19 g, 91%) was isolated in the form of a colorless syrup: $[\alpha]_D = +14.2$ (c, 1.42 in CHCl₃); R_f 0.55 (2:1 toluene–Me₂CO). ν_{max} (film): 3090, 3010, 2950, 2930, 2900, 1620, 1410, 1220, 1130, 1110, 1000, 960 cm⁻¹; ¹H NMR (CDCl₃): δ 2.27 (m, 1H, $J_{3a,3b}$ =14.6 Hz, $J_{2,3a}$ = 3.7 Hz, H-3a), 2.35 (ddd, 1H, J_{2,3b}=9.2 Hz, J_{3b,4}=4.9 Hz, H-3b), 3.76 (m, 1H, $J_{5,6a}=2.1$ Hz, $J_{5a,6b}=1.0$ Hz, H-5), 3.82-4.08 (m, 5H, 2×CH₂-dioxolane, and H-2), 4.13 (dd, 1H, *J*_{6a,6b}=13.1 Hz, H-6a), 4.48 (m, 2H, H-4 and H-6b), 5.12 (d, 1H, J_{1.2}=7.0 Hz, H-1), 5.45 (PhCH), 7.25–7.45 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 35.46 (C-3), 65.22 and 65.29 (2×CH₂-dioxolane), 67.17 (C-6), 75.36 (C-5), 76.44 (C-4), 79.57 (C-2), 99.81 (PhCH), 104.97 (C-1), 126.34, 128.31, 129.01 and 138.24 (Ph). CI-MS: *m*/*z* 279 (M⁺+1). HR-MS: Calcd for C₁₅H₁₈O₅: 278.1154. Found: *m*/*z* 278.1149.

2,5-Anhydro-6-O-benzoyl-4-bromo-3,4-dideoxy-L-ribohexose ethylene acetal (18). A mixture of 17 (0.097 g, 0.35 mmol), NBS (0.073 g, 0.41 mmol) and BaCO₃ (0.073 g, 0.41 mmol) in dry CCl₄ (5 mL) was refluxed in an atmosphere of N₂ for 1.5 h. The solvents were evaporated and the residue purified by column chromatography (17 g; 5:1 light petroleum–EtOAc) to yield pure 18 (0.079 g, 63%)as a colorless syrup: $[\alpha]_D = -26.8 (c, 0.7 \text{ in CHCl}_3); R_f 0.79$ (9:1 light petroleum-EtOAc). ν_{max} (film): 3080, 2960, 2900, 1725, 1600, 1460, 1275, 1100, 950 cm⁻¹; ¹H NMR (CDCl₃): δ 2.36 (ddd, 1H, $J_{3a,3b=}$ 13.4 Hz, $J_{2,3a}$ =7.4 Hz, $J_{3a,4}=5.9$ Hz, H-3a), 2.55 (ddd, 1H, $J_{3b,4}=6.9$ Hz, $J_{2,3b}=$ 6.7 Hz, H-3b), 3.28–4.04 (m, 4H, 2×CH₂-dioxolane), 4.28–4.55 (m, 5H, H-2, H-4, H-5 and 2×H-6), 4.93 (d, 1H, $J_{1,2}$ =3.4 Hz, H-1), 7.39–8.10 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 37.16 (C-3), 45.09 (C-4), 63.94 (C-6), 65.42 and 65.51 (2×*C*H₂-dioxolane), 78.81 (C-2), 85.38 (C-5), 103.76 (C-5), 103.76 (C-1), 128.36, 129.73 and 133.13 (Ph), 166.2 (C=O). CI-MS: m/z 357 (M⁺+1). HR-MS: Calcd for C₁₅H₁₇BrO₅: 356.0259. Found: *m*/*z* 356.0266.

2,5-Anhydro-4-*O***-benzoyl-6-bromo-6-deoxy-L-idose ethylene acetal (19).** A mixture of **13a** (0.10 g, 0.34 mmol) and BaCO₃ (0.04 g, 0.2 mmol) in dry CCl₄ (5 mL) was treated with NBS (0.073 g, 0.41 mmol) as described above. After 0.5 h the mixture was evaporated and the residue purified by

flash chromatography (49:1 CH₂Cl₂–Me₂–CO) to give pure **19** (0.084 g, 66%) as a solid. Recrystallization from CH₂Cl₂-hexane gave an analytical sample **19**: mp 142-143°C; $[\alpha]_{\rm D} = +44.6$ (c, 0.24 in CHCl₃); $R_{\rm f}$ 0.74 (1:1 cyclohexane-Me₂CO). v_{max} (KBr): 3430 (broad), 2980, 2920, 1730, 1610, 1290, 1130, 1090, 910 cm⁻¹; ¹H NMR (CDCl₃): δ 3.48–3.63 (m, 2H, $J_{6a,6b}$ =10.2 Hz, $J_{5.6a}$ = 8.5 Hz, J_{5.6b}=6.3 Hz, H-6a and H-6b), 3.85-4.11 (m, 4H, $2 \times CH_2$ -dioxolane), 4.18 (dd, 1H, $J_{1,2}=5.2$ Hz, $J_{2,3}=3.8$ Hz, H-2), 4.54 (dd, 1H, $J_{3,4}$ =1.2 Hz, H-3), 4.78 (ddd, 1H, J_{4.5}=3.4 Hz, H-5), 5.23 (d, 1H, H-1), 5.54 (dd, 1H, H-4), 7.41–8.06 (m, 5H, Ph); 13 C NMR (CDCl₃): δ 27.34 (C-6), 65.26 and 65.38 (2×CH2-dioxolane), 75.54 (C-3), 78.51 (C-4), 79.87 (C-5), 81.57 (C-2), 102.4 (C-1), 128.56, 129.1, 129.64 and 133.59 (Ph), 165.25 (C=O). CI-MS: m/z 373 (M⁺+1). Anal. Calcd for C₁₆H₁₉BrO₈: C 48.28, H 4.59. Found: C 48.00, H 4.32.

2,5-Anhydro-3-deoxy-L-*xylo***-hexose ethylene acetal (20).** (*Procedure A*) A solution of **16** (0.77 g, 3.95 mmol) and CF₃CO₂H (0.8 mL) in MeOH (7.2 mL) was stirred at room temperature for 0.5 h, and then evaporated by co-distillation with toluene. Flash chromatography (2:1 toluene–Me₂CO) of the residue gave pure **20** (0.505 g, 79%) as a colorless syrup.

Procedure B: a solution of 4,6-*O*-benzylidene derivative **17** (1.1 g, 3.95 mmol) in glacial acetic acid (6 mL) and water (1 mL) was stirred at reflux temperature for 7 h. The mixture was evaporated by co-distillation with toluene and the remaining crude **20** was purified by flash chromatography (EtOAc), to afford pure **20** (0.606 g, 81%) as a colorless syrup.

Procedure C: a solution of 15 (0.8 g, 2.9 mmol) in EtOH (20 mL) was hydrogenated over 10% Pd/C (0.3 g) for 4 h at room temperature. To the reaction mixture was then added glacial acetic acid (4 mL) and hydrogenation was continued for an additional 12 h. The mixture was filtered, the catalyst washed with EtOAc, and the combined filtrate and washings were evaporated by co-distillation with toluene. Flash column chromatography (EtOAc) of the residue afforded pure **20** (0.445 g, 83%) as a colorless syrup: $[\alpha]_{D} = +10.9$ (c, 1.34 in CHCl₃); $R_{\rm f}$ 0.19 (EtOAc). $\nu_{\rm max}$ (film): 3420 (broad), 2940, 2900, 1110–1050, 950 cm⁻¹; ¹H NMR (CDCl₃): δ 1.94 (ddd, 1H, $J_{3a,3b}$ =14.0 Hz, $J_{2,3a}$ =3.7 Hz, $J_{3a,4}=1.8$ Hz, H-3a), 2.32 (ddd, 1H, $J_{2,3b}=9.5$ Hz, $J_{3b,4}=$ 5.5 Hz, H-3b), 2.85 (bs, 1H, exchangeable with D_2O , OH), 3.78–4.11 (m, 8H, 2×CH₂-dioxolane, H-4, H-5 and 2×H-6), 4.24 (m, 1H, H-2), 5.0 (d, 1H, J_{1,2}=2.1 Hz, H-1), 5.30 (bs, 1H, exchangeable with D_2O , OH); ¹³C NMR (CDCl₃): δ 35.51 (C-3), 61.84 (C-6), 65.54 and 65.67 (2×CH₂-dioxolane), 71.98 (C-2), 77.6 (C-4), 83.29 (C-5), 103.84 (C-1). CI-MS: m/z 191 (M⁺+1). HR-MS: Calcd for C₈H₁₄O₅: 190.0841. Found: m/z 19.0838.

2,5-Anhydro-3-deoxy-6*O-p***-toluenesulfonyl-L***-xylo***-hexose ethylene acetal (21).** To a cooled $(-28^{\circ}C)$ solution of **20** (0.87 g, 4.57 mmol) in dry pyridine (8 mL) was added a cold $(-28^{\circ}C)$ solution of tosyl chloride (1.4 g, 7.34 mmol) in dry pyridine (8 mL). The reaction mixture was left at $-28^{\circ}C$ for 6 days, then poured into 6 M HCl (30 mL) and the resulting emulsion was extracted with dichloromethane (3×20 mL). The extracts were combined, washed with water, dried and evaporated to yellow oil. Flash column chromatography $(9:1 \text{ toluene} - \text{Me}_2\text{CO})$ yielded pure **21** (1.26 g, 80%) as a colorless syrup: $[\alpha]_{\rm D} = +15.4$ (c, 0.78 in CHCl₃); $R_{\rm f}$ 0.47 (EtOAc). ν_{max} (film): 3520–3400 (broad), 2930, 2900, 1600, 1360, 1200, 1120, 980 cm⁻¹, ¹H NMR (CDCl₃): δ 1.93 (d, 1H, $J_{3a,3b}$ =14.0 Hz, H-3a), 2.27 (ddd, 1H, $J_{2,3b}$ = 8.8 Hz, $J_{3b,4}$ =5.0 Hz, H-3b), 2.44 (s, 3H, MeC₆H₄SO₂), 3.73 (d, 1H, exchangeable with D_2O , $J_{4,OH}$ =10.7 Hz, OH), 3.83-4.07 (m, 5H, 2×CH₂-dioxolane, and H-5), 4.15 (m, 2H, $J_{6a,6b}$ =9.2 Hz, H-4 and H-6a), 4.31 (m, 2H, H-2 and H-6b), 4.95 (s, 1H, H-1), 7.33–7.8 (m, 4H, MeC₆H₄SO₂); ¹³C NMR (CDCl₃): δ 21.67 (*Me*C₆H₄SO₂), 34.75 (C-3), 65.58 and 65.66 (2×CH₂-dioxolane), 68.94 (C-6), 70.93 (C-4), 78.13 (C-2), 81.56 (C-5), 103.45 (C-1), 128.1, 129.82, 132.98 and 144.79 (MeC₆H₄SO₂). CI-MS: m/z 345 (M^++1) . HR-MS: Calcd for $C_{15}H_{20}O_7S$: 344.0930. Found: m/z 344.0939.

2,5-Anhydro-3,6-dideoxy-L-xylo-hexose ethylene acetal (22). A mixture of 21 (1.26 g, 3.66 mmol) and LiAlH₄ (0.7 g, 18.44 mmol) in dry THF (20 mL) was refluxed in an atmosphere of N₂ for 4 h. Excess of the reagent was decomposed by addition of EtOAc (1 mL), the mixture was filtered and the precipitate washed with EtOAc. The combined filtrate and washings were evaporated and the remaining crude 22 purified by flash chromatography (5:1 toluene-Me₂CO). Pure 22 (0.575 g, 90%) was thus obtained as a colorless syrup: $[\alpha]_{\rm D} = +43.4$ (c, 1.03 in CHCl₃); $R_{\rm f}$ 0.23 (5:1 toluene–Me₂CO). ν_{max} (film): 3460 (broad), 3000-2900, 1450, 1110, 1090, 980 cm⁻¹; ¹H NMR (CDCl₃): δ 1.25 (d, 3H, J_{5.6}=6.4 Hz, 3×H-6), 1.90 (d, 1H, $J_{3a,3b}=14.0$ Hz, H-3a), 2.26 (ddd, 1H, $J_{2,3b}=9.2$ Hz, $J_{3b,4}=$ 5.2 Hz, H-3b), 3.43 (d, 1H, exchangeable with D_2O , J_{4.OH}=11.0 Hz, OH), 3.76–4.11 (m, 6H, 2×CH₂-dioxolane, H-4, and H-5), 4.19 (d, 1H, H-2), 4.95 (s, 1H, H-1); ¹³C NMR (CDCl₃): δ 14.09 (C-6), 35.41 (C-3), 65.58 and 65.68 (2×CH₂-dioxolane), 72.31 (C-4), 77.04 (C-2), 80.19 (C-5), 103.78 (C-1). CI-MS: m/z 175 (M⁺+1). HR-MS: Calcd for C₈H₁₄O₄: 174.0892. Found: *m*/*z* 174.0885.

2,5-Anhydro-3,6-dideoxy-4-O-p-toluenesulfonyl-L-xylohexose ethylene acetal (23). To a solution of 22 (0.77 g, 4.42 mmol) in dry pyridine (10 mL) was added TsCl (2.5 g, 13.11 mmol) and the mixture was kept at room temperature for 48 h. The reaction mixture was poured into 6 M HCl (30 mL) and extracted with CH_2Cl_2 (4×15 mL). The combined extracts were washed with water, dried and evaporated to pale yellow oil. Flash chromatography (99:1 $CH_2Cl_2-Me_2CO$) of the residue gave pure 23 (1.165 g, 80%) as a colorless syrup: $[\alpha]_{D} = +16.8$ (c, 1.04 in CHCl₃); R_f 0.72 (EtOAc), R_f 0.55 (1:1 toluene-EtOAc). $\nu_{\rm max}$ (film): 3010–2910, 1610, 1370, 1200, 1120, 950 cm⁻¹; ¹H NMR (CDCl₃): δ 1.23 (d, 3H, $J_{5,6}$ =6.3 Hz, $3 \times H-6$), 2.1 (ddd, 1H, $J_{3a,3b}=15.0$ Hz, $J_{2,3a}=5.8$ Hz, $J_{3a,4}=$ 2.0 Hz, H-3a) 2.33 (ddd, 1H, $J_{2,3b}$ =8.6 Hz, $J_{3b,4}$ =6 Hz, H-3b), 2.45 (s, 3H, $MeC_6H_4SO_2$), 3.79 (m, 1H, $J_{1,2}=$ 6.2 Hz, H-2), 3.8-4.02 (m, 5H, 2×CH₂-dioxolane and H-5), 4.84 (d, 1H, H-1), 4.9 (m, 1H, J_{4.5}=3.7 Hz, H-4), 7.3-7.83 (m, 4H, MeC₆H₄SO₂); ¹³C NMR (CDCl₃): δ 14.47 (C-6), 21.73 (MeC₆H₄SO₂), 34.92 (C-3), 65.36 (2×CH₂dioxolane), 78.16 (C-2), 78.39 (C-5), 82.16 (C-4), 104.45 (C-1), 127.81, 129.93, 133.9 and 144.94 (MeC₆H₄SO₂).

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CI-MS: m/z 329 (M⁺+1). HR-MS: Calcd for C₁₅H₂₀O₆S: 328.0981. Found: m/z 328.0995.

2.5-Anhydro-4-O-benzoyl-3.6-dideoxy-L-ribo-hexose ethylene acetal (24). A mixture of 23 (1.15 g, 3.5 mmol) and KOBz (2.5 g, 15.65 mmol) in DMF (30 mL) was stirred at 100°C for 24 h. The solvent was removed by high vacuum distillation, the residue was treated with CH_2Cl_2 (2×20 mL), and the combined extracts were filtered and evaporated to yellow oil. Chromatographic purification on a column of flash silica (CH₂Cl₂) afforded pure 24 (0.64 g, 66%) as a colorless syrup: $[\alpha]_D = +0.5$ (c, 1.7 in CHCl₃); $R_f 0.7$ (19:1 CH₂Cl₂–Me₂CO), R_f 0.72 (1:1 toluene–EtOAc). v_{max} (film): 3000, 2950, 2900, 1730, 1610, 1460, 1390, 1290, 1130, 950 cm⁻¹; ¹H NMR (CDCl₃): δ 1.34 (d, 3H, $J_{5.6}=6.6$ Hz, 3×H-6), 2.15 (ddd, 1H, $J_{3a,3b}=13.8$ Hz, $J_{2,3a}=6.3$ Hz, $J_{3a,4}=2.4$ Hz, H-3a), 2.25 (ddd, 1H, $J_{3b,4}=$ 5.9 Hz, J_{2,3b}=9.4 Hz, H-3b), 3.82-4.07 (m, 4H, 2×CH₂dioxolane), 4.14–4.29 (m, 2H, $J_{1,2}$ =5.5 Hz, $J_{4,5}$ =2.7 Hz, H-2 and H-5), 4.93 (d, 1H, H-1), 5.15 (m, 1H, H-4), 7.4-8.06 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 19.78 (C-6), 32.97 (C-3), 65.43 and 65.56 (2×CH₂-dioxolane), 79.07 (C-2), 79.88 (C-4), 81.01 (C-5), 104.73 (C-1), 128.45, 129.65, 129.97 and 133.22 (Ph), 166.14 (C=O). CI-MS: m/z 279 (M^++1) . HR-MS: Calcd for C₁₅H₁₈O₅: 278.1154. Found: m/z 278.1149.

2,5-Anhydro-4-O-benzoyl-3,6-dideoxy-L-*ribo***-hexitol (26).** (*Procedure A*) A solution of **24** (0.64 g, 2.3 mmol) in a mixture of trifluoroacetic acid (7 mL) and 6 M HCl (1.75 mL) was kept at $+4^{\circ}$ C for 24 h. The reaction mixture was evaporated by co-distillation with toluene to unstable oil. Thus obtained crude **25** (0.7 g) was immediately dissolved in MeOH (15 mL) and reduced with NaBH₄ (0.2 g, 5.29 mmol) at room temperature for 2 h. The mixture was poured into saturated NaCl solution (15 mL) and extracted with CH₂Cl₂ (4×10 mL). The combined extracts were dried and evaporated to give crude **26**. Flash chromatography (4:1 toluene–EtOAc) of the residue afforded pure **26** (0.328 g, 60%) as a colorless syrup.

Procedure B: to a stirred and ice-cooled solution of alcohol 22 (0.7 g, 4.02 mmol), benzoic acid (0.95 g, 7.79 mmol) and triphenylphosphine (3.0 g, 11.44 mmol) in dry THF (40 mL) was added dropwise a solution of diethyl azodicarboxylate (3.5 mL, 22.22 mmol) in dry THF (15 mL). The mixture was stirred at 0°C for 15 min and then at room temperature for 20 h. The mixture was poured into saturated NaHCO₃ solution (100 mL) and extracted with CH₂Cl₂ (4×50 mL). The extract was dried and evaporated, and the residue was purified by flash chromatography (CH₂Cl₂), to give 24 (1.5 g) contaminated with a small amount of aromatic impurities. The impure sample 24 was dissolved in a mixture of trifluoroacetic acid (8 mL) and 6 M HCl (2 mL) and the solution was kept at $+4^{\circ}$ C for 24 h. After workup as described above (procedure A), the remaining crude aldehyde 25 was dissolved in MeOH (15 mL) and treated with NaBH₄ (0.25 g, 6.61 mmol) at room temperature for 2 h. After the workup according to procedure A, crude 26 was obtained and purified by column chromatography (50 g; 9:1 toluene–Me₂CO). Pure **26** (0.26 g, 27% from 22) was obtained as a colorless syrup: $[\alpha]_{\rm D} = -7.8$ (c, 1.09 in CHCl₃); $R_{\rm f}$ 0.27 (4:1 tolueneMe₂CO). ν_{max} (film): 3460 (broad), 2990, 2940, 2890, 1730, 1610, 1290, 1130 cm⁻¹; ¹H NMR (CDCl₃): δ 1.35 (d, 3H, $J_{5,6}$ =6.5 Hz, 3×H-6), 2.04 (ddd, 1H, $J_{3a,3b}$ = 13.7 Hz, $J_{2,3a}$ =5.6 Hz, $J_{3a,4}$ =1.9 Hz, H-3a), 2.24 (m, 2H, $J_{2,3b}$ =10.2 Hz, $J_{3b,4}$ =6.3 Hz, H-3b and OH), 3.59 (ddd, 1H, $J_{1a,1b}$ =11.9 Hz, $J_{1a,2}$ =4.6 Hz, H-1a), 3.87 (ddd, 1H, $J_{1b,2}$ =2.9 Hz, H-1b), 4.22 (dq, 1H, $J_{4,5}$ =2.6 Hz, H-5), 4.32 (m, 1H, H-2), 5.14 (dd, 1H, H-4), 7.4–8.08 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 19.80 (C-6), 33.04 (C-3), 63.80 (C-1), 79.06 (C-5), 80.36 (C-2), 80.64 (C-4), 128.37, 129.55, 129.87 and 133.16 (Ph), 166.11 (C=O). CI-MS: *m/z* 237 (M⁺+1). HR-MS: Calcd for C₁₃H₁₆O₄: 236.1049. Found: *m/z* 236.1041.

2,5-Anhydro-4-O-benzoyl-1-iodo-1,3,6-trideoxy-L-ribohexitol (27). To a solution of 26 (0.29 g, 1.23 mmol) in dry toluene (25 mL) were added successively imidazole (0.198 g, 2.24 mmol), Ph₃P (0.742 g, 2.83 mmol) and iodine (0.568 g, 2.24 mmol). The mixture was refluxed while stirring in an atmosphere of N_2 for 3 h, and then evaporated. Flash column chromatography (toluene) of the residue (1.8 g) yielded pure 27 (0.355 g, 84%) which was crystallized from hexane to give colorless needless: mp 68°C, $[\alpha]_{\rm D} = -7.6$ (c, 0.37 in CHCl₃); lit.⁷ mp 68°C, $[\alpha]_{\rm D} =$ -11.67 (c, 0.93 in CHCl₃); R_f 0.88 (4:1 toluene-Me₂CO). ¹H NMR (CDCl₃): δ 1.38 (d, 3H, $J_{5,6}$ =6.5 Hz, 3×H-6), 2.06 (ddd, 1H, $J_{3a,3b}$ =13.8 Hz, $J_{2,3a}$ =9.8 Hz, $J_{3a,4}$ =6.3 Hz, H-3a), 2.30 (ddd, 1H, $J_{2,3b}$ =5.4 Hz, $J_{3b,4}$ =1.7 Hz, H-3b), 3.3 (dd, 1H, $J_{1a,b}=10.2$ Hz, $J_{1a,2}=6.2$ Hz, H-1a), 3.37 (dd, 1H, J_{1b,2}=4.9 Hz, H-1b), 4.19 (m, 1H, H-2), 4.3 (dq, 1H, J_{4.5}=2.5 Hz, H-5), 5.16 (dt, 1H, H-4), 7.42-8.1 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 9.36 (C-1), 20.15 (C-6), 38.62 (C-3), 77.83 (C-2), 80.36 (C-4), 81.46 (C-5), 128.5, 129.68, 129.87 and 133.32 (Ph), 166.11 (C=O). CI-MS: m/z 347 $(M^++1).$

2,5-Anhydro-1-iodo-1,3,6-trideoxy-L-ribo-hexitol (28).To a solution of 27 (0.355 g, 1.02 mmol) in dry THF (5 mL) was added saturated methanolic K₂CO₃ solution (1 mL) and the suspension was stirred at room temperature for 1.5 h. The mixture was poured into saturated NaCl solution (15 mL), acidified with 6 M HCl, and extracted with CH_2Cl_2 (4×10 mL). The combined extracts were washed with brine, dried and evaporated. Column chromatography (50 g; 9:1 toluene-Me₂CO) of the residue (0.31 g) gave pure **28** (0.205 g, 83%) as a colorless syrup: $[\alpha]_D = -33.3$ (c, 0.88 in CHCl₃); lit.⁷ $[\alpha]_{\rm D} = -30.7$ (c, 0.87 in CHCl₃); $R_{\rm f}$ 0.38 (4:1 toluene–Me₂CO). ¹H NMR (CDCl₃): δ 1.23 (d, 3H, $J_{5,6}$ =6.4 Hz, 3×H-6), 1.87 (ddd, 1H, $J_{3a,3b}$ =13.3 Hz, $J_{2,3a}=8.7$ Hz, $J_{3a,4}=6.1$ Hz, H-3a), 2.03 (ddd, 1H, J_{2,3b}=2.9 Hz, J_{3b,4}=6.2 Hz, H-3b), 3.04 (bs, 1H, exchangeable with D₂O, OH), 3.21 (dd, 1H, $J_{1a,1b}$ =10.2 Hz, $J_{1a,2}$ = 6.1 Hz, H-1a), 3.28 (dd, 1H, $J_{1b,2}$ =4.8 Hz, H-1b), 3.95 (dq, 1H, $J_{4,5}$ =3.3 Hz, H-5), 4.02 (dt, 1H, H-4), 4.11 (m, 1H, H-2); ¹³C NMR (CDCl₃): δ 10.63 (C-1), 19.95 (C-6), 40.87 (C-3), 77.22 (C-2 and C-4), 83.31 (C-5). CI-MS: m/z 243 (M⁺+1).

(+)-Muscarine iodide (1). A sealed tube containing 28 (0.19 g, 0.78 mmol) and ethanolic 40% Me₃N (8 mL) was heated at 80°C for 3 h. The volatiles were evaporated and the syrupy residue partitioned between distilled water (4 mL) and EtOAc (3 mL). After removal of the aqueous

phase, the organic layer was washed with water (2 mL). The combined aqueous solutions were evaporated by co-distillation with toluene, to yield pure alkaloid **1** (0.22 g, 93%) as a pale yellow solid. (For ¹H and ¹³C NMR spectral data see Table 1). Recrystallization from 2-propanol afforded needles: mp 147–149°C, $[\alpha]_D$ =+7.6 (*c*, 0.4 in EtOH); lit.⁷ mp 149°C, $[\alpha]_D$ =+6.36 (*c*, 0.35 in EtOH).

2,5-Anhydro-4-O-benzoyl-3,6-dideoxy-L-xylo-hexose ethylene acetal (29). To a solution of 22 (0.162 g, 0.93) in dry pyridine (4 mL) was added benzoyl chloride (0.5 mL, 4.3 mmol). The mixture was kept at room temperature for 24 h, then acidified with 6 M HCl (12 mL) and extracted with CH_2Cl_2 (4×8 mL). The extracts were combined, washed successively with water and saturated NaHCO₃ solution, dried, and concentrated to an oil. Column chromatography on silica gel (40 g; 7:1 toluene-Me₂CO) afforded pure **29** (0.222 g, 86%) as a colorless syrup: $[\alpha]_{\rm D} = -6.7$ (c, 0.86 in CHCl₃); R_f 0.51 (4:1 toluene–Me₂CO). ν_{max} (film): 3000–2900, 1720, 1600, 1270, 1120, 940 cm⁻¹; ¹H NMR (CDCl₃): δ 1.34 (d, 3H, *J*_{5,6}=6.4 Hz, 3×H-6), 2.14 (ddd, 1H, $J_{3a,3b}=14.6$ Hz, $J_{3a,4}=1.8$ Hz, $J_{2,3a}=5.5$ Hz, H-3a), 2.56 (ddd, 1H, $J_{2,3b}$ =8.2 Hz, $J_{3b,4}$ =6.4 Hz, H-3b), 3.84–4.06 (m, 5H, 2×CH₂-dioxolane and H-2), 4.12 (m, 1H, $J_{4,5}$ 3.7H-5), 4.98 (d, 1H, $J_{1,2}$ =5.8 Hz, H-1), 5.48 (m, 1H, H-4), 7.4–8.14 (Ph); ¹³C NMR (CDCl₃): δ 14.47 (C-6), 34.91 (C-3), 65.37 and 65.39 (2×CH₂-dioxolane), 75.54 (C-4), 78.60 (C-2), 78.59 (C-5), 104.95 (C-1), 128.48, 129.69, 130.07 and 133.18 (Ph), 166.0 (C=O). CI-MS: m/z 557 (2M⁺+1), 279 (M⁺+1). HR-MS: Calcd for C₁₅H₁₈O₅: 278.1154. Found: *m*/*z* 278.1151.

2,5-Anhydro-4-*O***-benzoyl-3,6-dideoxy-L***-xylo***-hexitol (31).** (*Procedure A*) A solution of **29** (0.193 g, 0.69 mmol) in trifluoroacetic acid (2 mL) and 6 M HCl (0.5 mL) was stored at $+4^{\circ}$ C for 24 h. The mixture was evaporated by co-distillation with toluene, and the remaining crude aldehyde **30** (0.2 g) was reduced with NaBH₄ (0.06 g, 1.59 mmol) in MeOH (5 mL) following the procedure for preparation of **26**. Column chromatography (20 g; 4:1 toluene–Me₂CO) afforded pure **31** (0.097 g, 59%) as colorless syrup.

Procedure B: treatment of 22 (0.54 g, 3.1 mmol) with benzoyl chloride (1.5 mL, 12.91 mmol) in dry pyridine (6 mL) under the same reaction conditions as described above (preparation of 29) afforded crude 29 (1.6 g). Hydrolysis of the crude 29 with a mixture of trifluoroacetic acid (8 mL) and 6 M HCl (1 mL) at $+4^{\circ}$ C for 24 h gave the unstable aldehyde 30, which was subsequently reduced with NaBH₄ (0.5 g, 13.22 mmol) in MeOH (15 mL). After workup as described in the procedure for preparation of 26, followed by flash column chromatography (4:1 toluene- Me_2CO , pure **31** (0.385 g, 53%) was obtained as a colorless syrup: $[\alpha]_{D} = +25.2$ (c, 0.92 in CHCl₃); R_{f} 0.16 (4:1) toluene–Me₂CO). ν_{max} (film): 3450 (broad), 3000, 2960, 2890, 1740, 1610, 1460, 1300, 1130 cm⁻¹; ¹H NMR (CDCl₃): δ 1.33 (d, 3H, J_{5.6}=6.7 Hz, 3×H-6), 1.98 (ddd, 1H, $J_{3a,3b}=14.6$ Hz, $J_{2,3a}=6.1$ Hz, $J_{3a,4}=1.8$ Hz, H-3a), 2.17 (bs, 1H, exchangeable with D_2O , OH), 2.52 (ddd, 1H, $J_{2,3b}$ =8.5 Hz, $J_{3b,4}$ =6.5 Hz, H-3b), 3.65 (dd, 1H, $J_{1a,1b}$ 11.3 $J_{1a,2}$ =5.6 Hz, H-1a), 3.8 (dd, 1H, $J_{1b,2}$ =2.8 Hz, H-1b), 4.09 (m, 1H, J_{4.5}=3.9 Hz, H-5), 4.15 (m, 1H, H-2), 5.49 (m,

1H, H-4), 7.4–8.09 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 14.48 (C-6), 35.08 (C-3), 64.94 (C-1), 75.92 (C-4), 78.01 (C-2), 78.24 (C-5), 128.53, 129.62, 130.01, and 133.24 (Ph), 165.98 (C=O). CI-MS: *m*/*z* 237 (M⁺+1). HR-MS: Calcd for C₁₃H₁₆O₄: 236.1049. Found: *m*/*z* 236.1045.

2,5-Anhydro-4-O-benzoyl-1-iodo-1,3,6-trideoxy-L-xylohexitol (32). Treatment of 31 (0.359 g, 1.52 mmol) with imidazole (0.237 g, 3.48 mmol), Ph₃P (0.888 g, 3.38 mmol) and iodine (0.68 g, 2.68 mmol) in dry toluene (25 mL), according to the procedure described above for 27, yielded crude 32. Column chromatography on silica gel (150 g; 49:1 toluene-Me₂CO) afforded pure 32 (0.475 g, 90%) as a colorless syrup: $[\alpha]_D = +31.2$ (c, 0.96 in CHCl₃); $R_{\rm f}$ 0.74 (4:1 toluene–Me₂CO). $\nu_{\rm max}$ (film): 3000, 2950, 2870, 1730, 1610, 1290, 1120, 1190 cm⁻¹; ¹H NMR (CDCl₃): δ 1.34 (d, 3H, J_{5.6}=6.1 Hz, 3×H-6), 2.02 (ddd, 1H, $J_{3a,3b}$ =14.6 Hz, $J_{2,3a}$ =5.8 Hz, $J_{3a,4}$ =1.8 Hz, H-3a), 2.64 (ddd, 1H, $J_{2,3b}$ =8.2 Hz, $J_{3b,4}$ =6.4 Hz, H-3b), 3.31 (dd, 1H, $J_{1a,1b}$ 9.8 $J_{1a,2}$ =7.2 Hz, H-1a), 3.39 (dd, 1H, $J_{1b,2}$ =5.4 Hz, H-1b), 4.08–4.22 (m, 2H, J_{4.5}=4.0 Hz, H-2 and H-5), 5.51 (ddd, 1H, H-4), 7.42–8.11 (m, 5H, Ph); ¹³C NMR (CDCl₃): δ 9.18 (C-1), 14.75 (C-6), 39.34 (C-3), 75.68 (C-4), 77.51 (C-2), 78.92 (C-5), 128.5, 129.61, 129.89 and 133.22 (Ph), 165.87 (C=O). CI-MS: m/z 347 (M⁺+1). HR-MS: Calcd for C₁₃H₁₅IO₃: 346.0066. Found: *m*/*z* 346.0072.

2,5-Anhydro-1-iodo-1,3,6-trideoxy-L-xylo-hexitol (33). A solution of 32 (0.44 g, 1.27 mmol) in dry THF (5 mL) was treated with saturated methanolic K₂CO₃ solution (1 mL), as described above (procedure for 28), to afford crude 33. Flash chromatography (9:1 toluene-Me₂CO) yielded pure 33 (0.201 g, 65%) as an oil. Crystallization from hexane gave colorless needless: mp 63.5°C, $[\alpha]_D = -1.5$ (c, 1.13 in CHCl₃); lit.⁸ mp 62°C, $[\alpha]_{\rm p} = -0.34$ (c, 0.16 in CHCl₃); $R_{\rm f}$ 0.4 (4:1 toluene–Me₂CO). ¹H NMR (CDCl₃): δ 1.28 (d, 3H, $J_{5.6}=6.4$ Hz, 3×H-6), 1.77 (ddd, 1H, $J_{3a,3b}$ 14.3 $J_{2,3a}=5.2$ Hz, $J_{3a,4}=1.3$ Hz, H-3a), 1.97 (bs, 1H, exchangeable with D₂O, OH), 2.4 (ddd, 1H, *J*_{2,3b}=8.4 Hz, *J*_{3b,4}=6.0 Hz, H-3b), 3.31 (dd, 1H, $J_{1a,1b}$ 10.0 $J_{1a,2}$ =4.7 Hz, H-1a), 3.4 (dd, 1H, $J_{1b,2}$ =6.3 Hz, H-1b), 3.86 (m, 1H, $J_{4,5}$ =3.4 Hz, H-5), 3.94 (m, 1H, H-2), 4.17 (bs, 1H, H-4); ¹³C NMR (CDCl₃): δ 11.69 (C-1), 14.07 (C-6), 41.34 (C-3), 73.27 (C-4), 76.69 (C-2), 79.82 (C-5), CI-MS: m/z 243 (M⁺+1).

(+)-*epi*-Muscarine iodide (2). Iodo alcohol 33 (0.16 g, 0.66 mmol) was treated with a 40% ethanolic solution of Me₃N (15 mL) according to the same procedure as described for 1. The usual workup gave pure alkaloid 2 (0.19 g, 95%) as a yellow syrup. (For ¹H and ¹³C spectral data see Table 1). Crystallization from 2-propanol gave pale yellow needles: mp 172–173°C, $[\alpha]_D$ =+31.7 (*c*, 0.4 in H₂O); lit.⁸ mp 175°C, $[\alpha]_D$ =+32 (*c*, 0.55 in H₂O).

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